

# Maximizing Hydrogen Production by Cyanobacteria

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When incubated anaerobically, in the light, in the presence of  $C_2H_2$  and high concentrations of  $H_2$ , both Mo-grown *Anabaena variabilis* and either Mo- or V-grown *Anabaena azotica* produce large amounts of  $H_2$  in addition to the  $H_2$  initially added. In contrast,  $C_2H_2$ -reduction is diminished under these conditions. The additional  $H_2$ -production mainly originates from nitrogenase with the V-enzyme being more effective than the Mo-protein. This enhanced  $H_2$ -production in the presence of added  $H_2$  and  $C_2H_2$  should be of interest in approaches to commercially exploit solar energy conversion by cyanobacterial photosynthesis for the generation of molecular hydrogen as a clean energy source.

**Key words:** Hydrogenases, Alternative Nitrogenases, Photobiological Hydrogen Production in Cyanobacteria

## Introduction

Many cyanobacteria have the simplest nutrient requirements among all organisms. They thrive photoautotrophically solely in inorganic media, and  $N_2$ -fixing species do not even require combined nitrogen (nitrate, ammonia or organic nitrogen) for growth. Another characteristic feature of cyanobacteria is the ability to produce or consume molecular hydrogen. They possess two different enzyme complexes systems capable of catalyzing  $H_2$ -production. The first system, the nitrogenases, evolves  $H_2$  in parallel with  $NH_4^+$ -formation (Newton, 2007). Of the three known closely related types of nitrogenases (the Mo-, V- and Fe-only enzymes), both the Mo- and the V-nitrogenase have been reported for cyanobacteria (Kentemich *et al.*, 1988; Thiel, 1993; Boison *et al.*, 2006). The V-nitrogenase is less effective than the Mo-enzyme in catalyzing the reduction of both  $N_2$  and  $C_2H_2$  but consequently produces more  $H_2$  (Pau, 1991). The second system comprises the hydrogenases that catalyze the heterolytic cleavage of  $H_2$  into  $2 H^+$  and  $2 e^-$ . Cyanobacteria express two different Ni-containing hydrogenases (Tamagnini *et al.*, 2002; Schütz *et al.*, 2004): The uptake hydrogenase encoded by *hupL* and *hupS* is believed to be (mainly or exclusively) confined to heterocysts where it recycles the electrons lost as  $H_2$  during the  $N_2$ -fixation process. Because it feeds electrons into the respiratory complex II (Eisbrenner and Bothe, 1979),  $H_2$ -uptake by this enzyme is physiologically

unidirectional on thermodynamic grounds. The other enzyme, the bidirectional (reversible) hydrogenase, is NAD(P)H-dependent and is encoded by the *hoxEFUYH* gene cluster. Although its location in the cells has not been finally resolved, it might also function in the uptake of  $H_2$  under physiological conditions. This enzyme can, however, catalyze a transitory outburst of  $H_2$ -production when cells are exposed to an excess of reductant, *e.g.* under sudden high light intensities.

Over the years, many attempts have been made to couple photosynthetic electron transport with  $H_2$ -producing enzymes, either hydrogenase or nitrogenase. Solar energy conversion to  $H_2$  by such photosynthetic organisms would mean the generation of “clean” combustible energy. Due to the demand for alternative energy sources, there is a renewed interest in such systems at present (Pinto *et al.*, 2002; Tamagnini *et al.*, 2002; Cournac *et al.*, 2004; Levin *et al.*, 2004; Schütz *et al.*, 2004; Vignais and Colbeau, 2004; Shestakov and Mikheeva, 2006). Because  $H_2$  allows one to store large amounts of energy within small volumes,  $H_2$  is the energy carrier of choice for the future.  $H_2$ -production by cyanobacteria has, however, not reached a commercially acceptable level, because the efficiency in the conversion of incident light to  $H_2$  is only 1–2% (Yoon *et al.*, 2006) or even lower in many publications, although claims for a solar energy conversion factor of some 7% have been made in the past (Mitsui and Kumazawa, 1977).

Also problematic is the prerequisite for commercial H<sub>2</sub>-exploitation that sustained H<sub>2</sub>-production be achieved on a long-time scale. In addition, the clean up of the cyanobacterial culture after the termination of H<sub>2</sub>-production might not be easily achieved. A significant enhancement in H<sub>2</sub>-production by cyanobacteria must accrue if the cost-benefit ratio is to become promising.

An observation made by us during laboratory student course experiments may help to enhance the cyanobacterial H<sub>2</sub>-production rates. C<sub>2</sub>H<sub>2</sub>-reduction by heterocystous cyanobacteria is more resistant to exposure to O<sub>2</sub> when the cells are incubated with H<sub>2</sub>. H<sub>2</sub>-consumption by hydrogenases in a Knallgas-type reaction apparently removes O<sub>2</sub> from the vicinity of nitrogenase thereby preventing its denaturation by O<sub>2</sub> (Bothe *et al.*, 1977). In experiments designed to demonstrate this effect, we were surprised to find that the control assays with no O<sub>2</sub> present in the vessels had significantly higher H<sub>2</sub>-concentrations at the end of the experiments than was initially added. This anaerobic H<sub>2</sub>-production in intact cells in the presence of both H<sub>2</sub> and C<sub>2</sub>H<sub>2</sub> will be characterized here for the two heterocystous cyanobacteria *Anabaena variabilis* and *Anabaena azotica* where the latter of which was grown with either Mo or V in the medium.

## Materials and Methods

### *Organisms and their growth*

*Anabaena variabilis* ATTC 29413 was purchased. *Anabaena azotica* FACHB-118 from the Freshwater Algae Culture Collection of the Institute of Hydrobiology, Chinese Academy of Sciences, Wuhan, Hubei, China was kindly supplied by Professor Dai Heping, Wuhan, P. R. China. The cyanobacteria were grown in BG11 medium without combined nitrogen (Rippka *et al.*, 1979) at 25 °C under continuous gassing with a mixture of air/CO<sub>2</sub> and with continuous illumination (light intensity approximately at the surface of the vessels 70  $\mu\text{E m}^{-2} \text{ s}^{-1}$ ) as described by Neuer and Bothe (1985). To induce the expression of V-nitrogenase, no Mo but 10  $\mu\text{M}$  V<sub>2</sub>O<sub>5</sub> was added to the medium. No attempt was made to remove Mo from the water or from any other chemical.

Mo-grown *A. variabilis* had a heterocyst frequency of  $(5.2 \pm 0.7)\%$  ( $n = 18$ , 100 cells counted for each). The number of heterocysts in the filaments was higher with *A. azotica*; it averaged

$(8.7 \pm 1.0)\%$  ( $n = 9$ ) for Mo- and  $(11.5 \pm 1.1)\%$  ( $n = 12$ ) for V-grown cultures. *A. azotica* produced significantly more extracellular polysaccharides and had a two- to threefold lower growth rate than *A. variabilis*. The growth rate of *A. azotica* was not impaired by substituting Mo by V in the medium.

### *Activity measurements by gas chromatography*

Cyanobacterial filaments were concentrated in the culture tubes by their rapid self-sedimentation after terminating gassing. The experiments were performed in 7.0 ml Fernbach flasks covered with gas-tight suba seals into which C<sub>2</sub>H<sub>2</sub> (1 ml) and H<sub>2</sub> (amounts given in the abscissa of the figures) were injected with a syringe. Prior to these injections of both H<sub>2</sub> and C<sub>2</sub>H<sub>2</sub>, the vessels were made anaerobic by gassing with argon. The dashed or dotted part of the columns in Fig. 1 indicates the amount of H<sub>2</sub> injected into the vessels immediately determined after the injection of the gases. Because H<sub>2</sub> had to be injected against the argon pressure in the vessels, H<sub>2</sub> did not proportionally increase with the milliliters of H<sub>2</sub> added by the syringe. The term "complete" in the figures means that argon was substituted by H<sub>2</sub> during the gassing procedure. The gas phase in the vessels then had a pressure of approximately 1 bar of H<sub>2</sub> that, however, did not prevent a further generation of H<sub>2</sub> by the cells against this pressure. The experiments were performed with 3 ml of cyanobacterial suspensions [total amount 0.035 mg chlorophyll, determined as described by Biggins (1967)] with the Fernbach flasks rotating on a horizontal shaker at 20 °C and a light intensity at the vessels of about 300  $\mu\text{E m}^{-2} \text{ s}^{-1}$ .

H<sub>2</sub> was quantified by gas chromatography using a Perkin Elmer 8500 gas chromatograph equipped with a thermal conductivity detector and a molecular sieve 5 Å column (45/60 mesh, 2 m  $\times$  1/8 inch). For C<sub>2</sub>H<sub>2</sub>-reduction, the C<sub>2</sub>H<sub>4</sub> formed was also determined by gas chromatography using a flame ionization detector and a Carbosieve SII column (100/200 mesh, 10 feet  $\times$  1/8 inch). For further details see Kentemich *et al.* (1988).

### *Reproducibility of the experiments*

All experiments presented in the figures and tables have been performed at least three times with the same outcome. However, the specific activities were variable from experiment to experiment, as known for physiological assays. The specific activi-

ties depend on the physiological state of the cells (batch cultures have been used; cells had to be concentrated prior to the start of the experiments which could hardly be standardized; the time needed for preparing the assay was variable). The content of cyanobacterial cells in the vessels varied by a factor of 1.5–2 from one experiment to the next. Therefore only representative data can be given here as in all other previous publications from this laboratory on this subject.

## Results

When *A. variabilis* or *A. azotica*, either Mo- or V-grown, were incubated anaerobically in the light in the presence of high concentrations of both  $H_2$  and  $C_2H_2$  for 4 h, cells formed significantly more  $H_2$  than was added at the start of the experiments.

Rates of this additional  $H_2$ -formation were roughly ten times higher than the  $C_2H_2$ -reduction activities (Table I). The greater the amount of  $H_2$  initially added to the assay vessels, the greater was the amount of additional  $H_2$  produced. This was the case for both V- and Mo-grown *A. azotica* and also for Mo-grown *A. variabilis* at the higher  $H_2$ -concentrations (Fig. 1). This formation of additional  $H_2$  proceeded linearly for at least 2 h (Fig. 2 for Mo- and V-grown *A. azotica*, same result for Mo-grown *A. variabilis*, not shown). The formation of additional  $H_2$  was more obvious when the assays were performed at lower light intensities and with smaller amounts of the cells in the reaction vessels, as in the current experiments (comparative data not shown). Without adding  $H_2$  to the vessels, net  $H_2$ -production was less than

Assay condition	$C_2H_2$ -reduction [ $\mu\text{mol } C_2H_4$ formed/ h · mg chlorophyll]	$H_2$ -formation [ $\mu\text{mol } H_2$ formed/ h · mg chlorophyll]
<i>Anabaena azotica</i>		
V-culture, without $H_2$	3.4	0.6
V-culture, plus $H_2$	3.3	40
Mo-culture, without $H_2$	4.0	0.1
Mo-culture, plus $H_2$	1.8	9
<i>Anabaena variabilis</i>		
Mo-culture, without $H_2$	7.8	0.5
Mo-culture, plus $H_2$	3.3	24

Table I.  $C_2H_2$ -reduction and  $H_2$ -formation activities by *Anabaena azotica* and *A. variabilis*.

The experiments were performed under argon in 7.0 ml Fernbach flasks into which  $C_2H_2$  and  $H_2$  (each 1 ml) were injected.  $C_2H_4$ -formation was determined after 1 h and  $H_2$ -production after 4 h. For other details see Materials and Methods.

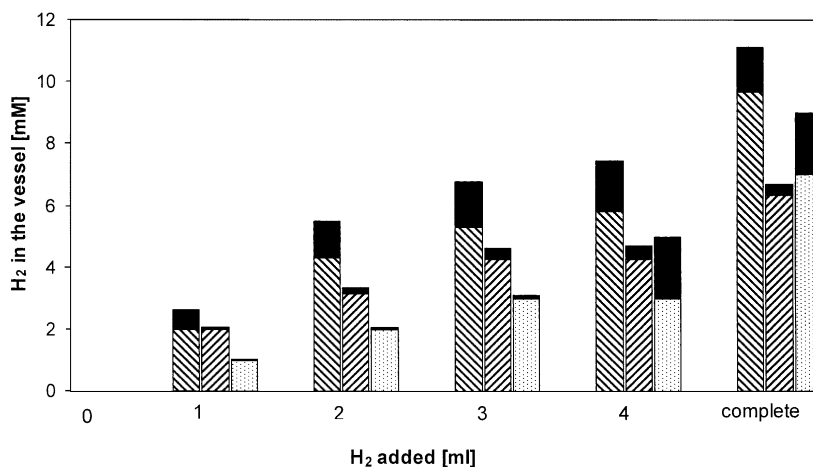


Fig. 1.  $H_2$ -production by *Anabaena azotica* (V- or Mo-grown) and *A. variabilis*. The black part of the columns indicates the concentration of  $H_2$  formed per 4 h and mg chlorophyll in the vessels, the dashed or dotted part of the columns the amount of  $H_2$  added to the vessels by syringes and determined by gas chromatography at the start of the experiments. Columns with dashes to the left, *A. azotica* V-grown; columns with dashes to the right, *A. azotica* Mo-grown; columns with dots, *A. variabilis* Mo-grown. Complete means gas phase  $H_2$  (about 1 bar).

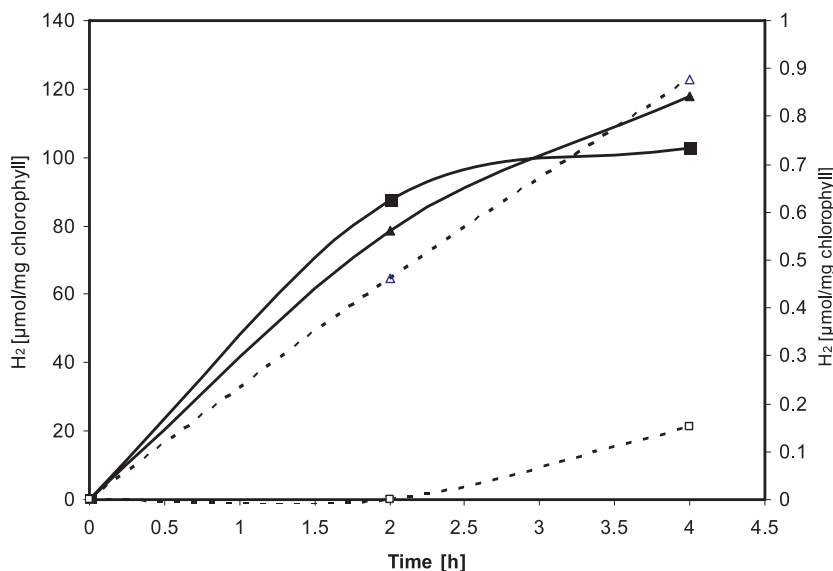


Fig. 2. Kinetics of H<sub>2</sub>-formation by Mo- and V-grown *Anabaena azotica*. Solid symbols and lines indicate assays in Fernbach flasks supplemented with 1 ml H<sub>2</sub>; the appropriate rate scale is shown on the left ordinate. Dashed lines and open symbols indicate assays without H<sub>2</sub> supplementation; the appropriate rate scale is shown on the right ordinate. In this experiment, H<sub>2</sub>-production was atypically as high in Mo- as in V-grown cultures which happened with approximately one out of ten cultures. Triangles, Mo-grown culture; squares, V-grown culture.

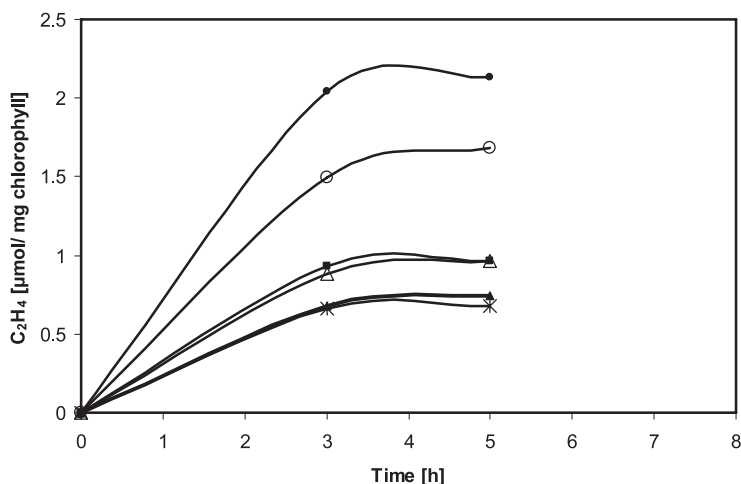


Fig. 3. Inhibition of C<sub>2</sub>H<sub>2</sub>-reduction by increasing concentrations of H<sub>2</sub> added to the assays, using Mo-grown *A. azotica*. Top line, no H<sub>2</sub> injected into the vessels; the following lines, 1, 2, 3, and 4 ml H<sub>2</sub> injected; the bottom line, 100% H<sub>2</sub> gas phase. The inhibition pattern was the same for V-grown *A. azotica* and for Mo-grown *A. variabilis* (not documented).

1.0 μmol/h · mg chlorophyll (Table I and right ordinate of Fig. 2), in accordance with earlier determinations (Bothe *et al.*, 1978).

In the same experiments, C<sub>2</sub>H<sub>4</sub>-formation showed a steady decline as increasing amounts of H<sub>2</sub> were added (Fig. 3) and stabilized at about

30% of the original H<sub>2</sub>-free activity under 100% H<sub>2</sub>. A concentration of 6 mM H<sub>2</sub> in the assays decreased C<sub>2</sub>H<sub>4</sub>-formation by about 50%, irrespective of whether the cells were Mo- or V-grown.

Both H<sub>2</sub>-formation (in addition to the H<sub>2</sub> added) and C<sub>2</sub>H<sub>2</sub>-reduction were greatly stimu-

Assay condition	C <sub>2</sub> H <sub>4</sub> -formation [ $\mu\text{mol/h} \cdot \text{mg}$ chlorophyll]	H <sub>2</sub> -formation ( $\mu\text{mol/h} \cdot \text{mg}$ chlorophyll)
Control without H <sub>2</sub>	2.4	0.1
H <sub>2</sub> alone	1.2	34.5
H <sub>2</sub> , 40 $\mu\text{M}$ DCMU	0.5	13.2
H <sub>2</sub> , 80 $\mu\text{M}$ DCMU	0.4	11.3
H <sub>2</sub> , 6 $\mu\text{M}$ CCP	1.0	17.9
H <sub>2</sub> , 12 $\mu\text{M}$ CCP	0.8	15.8
H <sub>2</sub> , 6 $\mu\text{M}$ FCCP	0.3	12.4
H <sub>2</sub> , 12 $\mu\text{M}$ FCCP	0.2	9.0
H <sub>2</sub> , vessels kept in darkness	0.0	6.0
H <sub>2</sub> , N <sub>2</sub> instead of C <sub>2</sub> H <sub>2</sub>	–	0.0

Table II. C<sub>2</sub>H<sub>4</sub>- and H<sub>2</sub>-formations by vanadium-grown *Anabaena azotica*: Effects of inhibitors.

The experimental conditions were as for Table I but 3 ml H<sub>2</sub> were injected to the vessels this case.

DCMU, dichlorophenyl dimethyl urea, first dissolved in 25% methanol, then diluted tenfold with water to give the final concentration in the vessel; CCP, carbonyl-cyanide-phenylhydrazone; FCCP, carbonyl-cyanide-*p*-trifluoromethoxy-phenylhydrazone (*p*-CF<sub>3</sub>-CCP).

lated by light and were sensitive to inhibitors of photosynthetic electron transport (Table II). The herbicide DCMU (dichlorophenyl dimethyl urea), which blocks at the acceptor side of photosystem II, affected both additional H<sub>2</sub>-formation and C<sub>2</sub>H<sub>2</sub>-reduction to approx. the same extent. Uncouplers also severely affected both activities with FCCP (carbonyl-cyanide-phenylhydrazone) being more effective than CCP (carbonyl-cyanide-*p*-trifluoromethoxyphenyl-hydrazone). Because the activities of nitrogenases, but not of hydrogenases, are energy (ATP)-dependent, the additional H<sub>2</sub>-production must have been produced mainly by nitrogenase. No additional H<sub>2</sub>-formation was observed when N<sub>2</sub> replaced C<sub>2</sub>H<sub>2</sub> as the nitrogenase substrate (Table II). H<sub>2</sub>-production was also not seen in nitrate-grown *A. variabilis* when nitrogenase was not expressed (not documented).

## Discussion

In the presence of N<sub>2</sub>, cells might be in balance with respect to their C/N ratio, and this balance is not affected by the addition of H<sub>2</sub>. The H<sub>2</sub>-gas formed in parallel with ammonia production in nitrogenase catalysis is immediately recycled by hydrogenases. This is manifested by rather little net production in intact cyanobacterial cells. In the presence of C<sub>2</sub>H<sub>2</sub>, however, high concentrations of H<sub>2</sub> cause a disturbance of this C/N ratio, possibly due to the fact that reductants generated photosynthetically cannot be properly utilized anymore to meet the N-demand of the cells. C<sub>2</sub>H<sub>2</sub>-reduction is diminished, and the excess of reductants is disposed of to reduce protons. This inhibition of C<sub>2</sub>H<sub>2</sub>-reduction by H<sub>2</sub> and the parallel production of additional H<sub>2</sub> in the presence of high concentration of H<sub>2</sub> in the assays are apparently hitherto

unreported observations. It has, however, been reported for bacteroids that high concentrations of H<sub>2</sub> inhibit N<sub>2</sub>-fixation, and that H<sub>2</sub>-recycling by hydrogenase prevents the build-up of high inhibitory concentrations of H<sub>2</sub> formed during N<sub>2</sub>-reduction in nodules (Dixon, 1972).

The high concentrations of H<sub>2</sub> and C<sub>2</sub>H<sub>2</sub> unlikely affect nitrogenase itself. Hydrogen is a competitive inhibitor of N<sub>2</sub>-reduction but not in the reduction of all other nitrogenase substrates including C<sub>2</sub>H<sub>2</sub> (Burns and Hardy, 1975). In cyanobacteria, C<sub>2</sub>H<sub>2</sub> (and CO) was described to inhibit the photosystem I or respiration-dependent uptake of H<sub>2</sub> (Bothe *et al.*, 1977) which could partly explain the enhanced H<sub>2</sub>-production. However, this does not explain the effect of the high concentrations of H<sub>2</sub>. The primary target of these high concentrations of H<sub>2</sub> remains obscure and speculative. In line with the present findings, a two- to fourfold increase of light-induced H<sub>2</sub>-evolution was described for the cyanobacterium *Nostoc muscorum* preincubated under hydrogen and argon years ago. The parallel inhibition was, however, not detected (Scherer *et al.*, 1980).

The light stimulation and the results with the inhibitors indicate that reducing equivalents for H<sub>2</sub>-formation are produced mainly photosynthetically. The reductants generated in the light might be disposed by H<sub>2</sub>-production via ferredoxin and nitrogenase to avoid an overreduction in the cells and an imbalance in the C/N ratio. A direct coupling of H<sub>2</sub>-production with reduced ferredoxin and hydrogenase(s) can be ruled out, since both cyanobacterial hydrogenases (uptake and bidirectional enzyme) do not utilize this electron carrier (Tamagnini *et al.*, 2002; Schütz *et al.*, 2004). It could be assumed that H<sub>2</sub>-production comes from ferre-



doxin, FNR [NAD(P)H:ferredoxin oxidoreductase], NAD(P)<sup>+</sup> and bidirectional hydrogenase. Such an activity should, however, be insensitive to uncouplers. Such an interpretation can also not explain that H<sub>2</sub>-formation is higher in V- than in Mo-grown cells. To a minor extent, reducing equivalents for H<sub>2</sub>-production may also come from carbohydrate metabolism, since the inhibitors of photosynthesis do not completely block the H<sub>2</sub>-production. Reductants generated during fermentative degradations of reserve material may be disposed of via pyruvate:ferredoxin oxidoreductase, FNR and bidirectional hydrogenase. Cyanobacteria are known to rapidly switch over to fermentation when exposed to the dark (Stal and Mozelaar, 1997). This smaller part of the H<sub>2</sub>-formation would then be produced by hydrogenase(s) and not by nitrogenase.

As first demonstrated for *Azotobacter vinelandii*, alternative nitrogenases produce more H<sub>2</sub> than the Mo-enzyme (Bishop and Joerger, 1990). This is also seen in the cyanobacteria *A. variabilis* (Kentemich *et al.*, 1988) and *A. azotica* (present study) having expressed the V-nitrogenase. Compared with the V-enzyme, an even higher H<sub>2</sub>-formation and a lower C<sub>2</sub>H<sub>2</sub>-reduction rate is seen with the Fe-only nitrogenase (Bishop and Joerger, 1990). Up to this day, an Fe-only nitrogenase has not been described for cyanobacteria, and the V-nitrogenase is known only for few isolates. An Fe-only nitrogenase is known for the photosynthetic bacterium *Rhodobacter capsulatus* (Newton, 2007) where it could drive photosynthetic H<sub>2</sub>-production, however not at the expense of H<sub>2</sub>O as the electron donor. The occurrence of an Fe-only nitrogenase in cyanobacteria cannot be ruled out, although it is not found in the genome of *A. variabilis* and *Anabaena* 7120. Negative results of growth experiments performed in the absence of both Mo and V in the medium are not convincing, since it is not easy to obtain both Mo- and V-deficiency in cultures due to the low demands of cells for these elements. Giving an example outside of cyanobacteria, the Fe-only nitrogenase has now, unexpectedly, been detected in the plant growth promoting bacterium *Azospirillum brasilense* Sp245 from the analysis of its completely sequenced genome (Zhulin, 2007).

For maximizing H<sub>2</sub>-production by cyanobacteria for potential applications, it is suggested to take *Anabaena variabilis* with an expressed V-nitroge-

nase. In contrast, *A. azotica* has higher amounts of heterocysts as the site of nitrogenase under aerobic growth conditions, but forms fairly large amounts of exopolysaccharides (slime) and grows slower than *A. variabilis*. Incubation of the cultures with high concentrations of H<sub>2</sub> and of C<sub>2</sub>H<sub>2</sub> enhances H<sub>2</sub>-formation, and this present observation can possibly be exploited for applications. For doing so, conditions still have to be optimized. Mutants defective in hydrogenases to prevent recycling of H<sub>2</sub> might also augment the photosynthetically driven H<sub>2</sub>-production by cyanobacteria. Indeed, some years ago, we observed higher H<sub>2</sub>-production rates in an *A. variabilis* mutant obtained by classical NTG mutagenesis which was affected in the recycling of H<sub>2</sub> produced by nitrogenase (Mikheeva *et al.*, 1995). Mutants defective in either uptake, bidirectional or both uptake and bidirectional hydrogenase have been constructed for *Anabaena* 7120, and activity measurements indicated that only the mutant defective in the uptake hydrogenase produced more H<sub>2</sub> (Masukawa *et al.*, 2002). Work with hydrogenase mutants from other cyanobacteria also showed increased H<sub>2</sub>-evolution rates than the wild-types (Happe *et al.*, 2001; Lindblad *et al.*, 2002; Cournac *et al.*, 2004; Yoshino *et al.*, 2007). A mutant in *A. variabilis* having expressed the V-nitrogenase would be better suited to optimize H<sub>2</sub>-production by working out the best culture conditions such as stability of the cultures for sustained gas production, best cell density or light intensity. The alternative approach to utilize a cyanobacterial hydrogenase will likely be less successful, since both enzymes mainly function in H<sub>2</sub>-utilization. Other attempts consist in engineering a foreign hydrogenase and in coupling it with photosystem I for H<sub>2</sub>-production. In such a recent approach, the membrane-bound enzyme from *Ralstonia eutropha* was coupled directly to a modified component (PsaE) of the acceptor side of the cyanobacterial photosystem I (Ihara *et al.*, 2006). The small H<sub>2</sub>-formation rates obtained with this artificial system may reflect that all such approaches are still far away from potential applications.

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- Biggins J. (1967), Photosynthetic reactions by lysed protoplasts and particle preparations from the blue-green alga, *Phormidium luridum*. *Plant Physiol.* **42**, 1447–1456.
- Bishop P. and Joerger R. D. (1990), Genetics and molecular biology of alternative nitrogen fixing systems. *Annu. Rev. Plant Physiol. Plant Mol. Biol.* **41**, 109–125.
- Boison G., Steingen C., Stal L. J., and Bothe H. (2006), The rice field cyanobacteria *Anabaena azotica* and *Anabaena* sp. CH1 express vanadium-dependent nitrogenase. *Arch. Microbiol.* **186**, 367–376.
- Bothe H., Tennigkeit J., Eisbrenner G., and Yates M. G. (1977), The hydrogenase-nitrogenase relationship in the blue-green alga *Anabaena cylindrica*. *Planta* **133**, 237–242.
- Bothe H., Distler E., and Eisbrenner G. (1978), Hydrogen metabolism in blue-green algae. *Biochimie* **60**, 277–289.
- Burns R. C. and Hardy R. W. F. (1975), Nitrogen Fixation in Bacteria and Higher Plants. Springer, Heidelberg, New York, p. 189.
- Cournac L., Guedeney G., Peltier G., and Vignais P. M. (2004), Sustained photoevolution of molecular hydrogen in a mutant of *Synechocystis* sp strain PCC 6803 deficient in the type I NADH-dehydrogenase complex. *J. Bacteriol.* **186**, 1737–1746.
- Dixon R. O. D. (1972), Hydrogenase in legume root nodule bacteroids, occurrence and properties. *Arch. Microbiol.* **85**, 193–201.
- Eisbrenner G. and Bothe H. (1979), Modes of electron transfer from molecular hydrogen in *Anabaena cylindrica*. *Arch. Microbiol.* **123**, 37–45.
- Happe T., Schütz K., and Böhme H. (2001), Transcriptional and mutational analysis of the uptake hydrogenase of the filamentous cyanobacterium *Anabaena variabilis*. *J. Bacteriol.* **182**, 1624–1631.
- Ihara M., Nishihara H., and Yoon K.-S. (2006), Light-driven production by a hybrid complex of a [NiFe]-hydrogenase and the cyanobacterial photosystem I. *Photochem. Photobiol.* **82**, 676–682.
- Kentemich T., Danneberg G., Hundeshagen B., and Bothe H. (1988), Evidence for the occurrence of the alternative, vanadium-containing nitrogenase in the cyanobacterium *Anabaena variabilis*. *FEMS Microbiol. Lett.* **51**, 19–24.
- Levin D. B., Pitt L., and Love M. (2004), Biohydrogen production: prospects and limitations to practical application. *Int. J. Hydrogen Energy* **29**, 173–185.
- Lindblad P., Christensson K., Lindberg P., Federov A., Pinto G., and Tsygankov A. (2002), Photoproduction of H<sub>2</sub> by wildtype *Anabaena* PCC 7120 and a hydrogenase deficient mutant: from laboratory experiments to outdoor culture. *Int. J. Hydrogen Energy* **27**, 1271–1281.
- Masukawa H., Mochimaru M., and Sakurai H. (2002), Disruption of the uptake hydrogenase gene, but not of the bidirectional hydrogenase gene, leads to enhanced photobiological hydrogen production by the nitrogen-fixing cyanobacterium *Anabaena* sp. PCC 7120. *Appl. Microbiol. Biotechnol.* **58**, 618–624.
- Mikheeva L. E., Schmitz O., Shestakov S. V., and Bothe H. (1995), Mutants of the cyanobacterium *Anabaena variabilis* altered in hydrogenase activities. *Z. Naturforsch.* **50c**, 505–510.
- Mitsui A. and Kumazawa S. (1977), Hydrogen production by marine photosynthetic organisms as potential energy resource. In: *Biological Solar Energy Conversion* (Mitsui A., Miyachi S., San Pietro A., and Tamura S., eds.). Academic Press, London, New York, San Francisco, pp. 275–289.
- Neuer G. and Bothe H. (1985), Electron donation to nitrogenase in heterocysts of cyanobacteria. *Arch. Microbiol.* **143**, 185–191.
- Newton W. E. (2007), Physiology, biochemistry, and molecular biology of nitrogen fixation. In: *Biology of the Nitrogen Cycle* (Bothe H., Ferguson S. J., and Newton W. E., eds.). Elsevier, Amsterdam, pp. 109–129.
- Pau R. N. (1991), The alternative nitrogenases. In: *Biology and Biochemistry of Nitrogen Fixation* (Dilworth M. J. and Glenn A. R., eds.). Elsevier, Amsterdam, pp. 37–57.
- Pinto F. A. L., Troshina O., and Lindblad P. (2002), A brief look at three decades of research on cyanobacterial hydrogen evolution. *Int. J. Hydrogen Energy* **27**, 1209–1215.
- Rippka R., Deruelles J., Waterbury J. B., Herdman M., and Stanier R. Y. (1979), Generic assignments, strain histories and properties of pure cultures of cyanobacteria. *J. Gen. Microbiol.* **111**, 1–61.
- Scherer S., Kerfin W., and Böger P. (1980), Increase of nitrogenase activity in the blue-green alga *Nostoc muscorum* (cyanobacterium). *J. Bacteriol.* **144**, 1017–1023.
- Schütz K., Happe T., Troshina O., Lindblad P., Leitao E., Oliveira P., and Tamagnini P. (2004), Cyanobacterial H<sub>2</sub>-production – a comparative analysis. *Planta* **218**, 350–359.
- Shestakov S. V. and Mikheeva L. E. (2006), Genetic control of hydrogen metabolism in cyanobacteria. *Russ. J. Genetics* **42**, 1272–1284.
- Stal L. J. and Mozelaar R. (1997), Fermentation in cyanobacteria. *FEMS Microbiol. Rev.* **21**, 179–211.
- Tamagnini P., Axelsson R., Lindberg P., Oxelfelt F., Wünschiers R., and Lindblad P. (2002), Hydrogenases and hydrogen metabolism of cyanobacteria. *Microbiol. Mol. Biol. Rev.* **66**, 1–20.
- Thiel T. (1993), Characterization of genes for an alternative nitrogenase in the cyanobacterium *Anabaena variabilis*. *J. Bacteriol.* **175**, 6276–6286.
- Vignais P. M. and Colbeau A. (2004), Molecular biology of microbial hydrogenases. *Curr. Issues Mol. Biol.* **6**, 159–188.
- Yoon J. H., Shin J. H., Kim M. S., Sim S. J., and Park T. H. (2006), Evaluation of conversion efficiency of light to hydrogen energy by *Anabaena variabilis*. *Int. J. Hydrogen Energy* **31**, 721–727.
- Yoshino F., Ikeda H., Masukawa H., and Sakurai H. (2007), High photobiological hydrogen production activity of a *Nostoc* sp PCC 7422 uptake hydrogenase-deficient mutant with high nitrogenase activity. *Marine Biotechnol.* **9**, 101–112.
- Zhulin B. (2007), First insights into the *Azospirillum* genome. In: *15th International Congress on Nitrogen Fixation*, Capetown, S. A. (in press).